**Supplementary Material for** 

### The Influence of the Framework Core Residues on the Biophysical Properties of Immunoglobulin Heavy Chain Variable Domains

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# Additional information about denaturant-induced denaturation of the scFv fragments studied

Tryptophan fluorescence as spectral probe for the unfolding of the scFv. The scFv fragments huV<sub>k</sub>3-huV<sub>H</sub>3, huV<sub>k</sub>3-huV<sub>H</sub>3•MocB and huV<sub>k</sub>3-huV<sub>H</sub>3•1 contain 5 Trp residues, a sixth fully buried Trp is present in the lower core of the V<sub>H</sub> domain of  $huV_{\kappa}3$ - $huV_{H}3$ •1 (Figure S1). Comparison of the spectra of the native and denatured isolated huV $\kappa$ 3 and huV<sub>H</sub>3 domains and of the scFv composed of these domains suggests that the conserved core Trp L43 and Trp H43 are highly quenched in the native domains (Jäger et al., 2001). Thus, the fluorescence intensity increases when the isolated  $huV\kappa3$  and  $huV_H3$  domains unfold. However, the specific fluorescence of the scFv is higher than the sum of the fluorescence of the two isolated domains. This is due to the contribution of the three  $V_{\rm H}$  tryptophan residues buried in the interface between the between the V<sub>L</sub> and the V<sub>H</sub> domain, Trp H54 (conserved), H109 (CDR-H3) and H139 (conserved). Their contribution is higher when they are buried in the domain interface (native scFv) than when these residues are solvent exposed (isolated domains). Upon domain dissociation, not only does the fluorescence quantum yield of these interface Trp residues decrease, but also the fluorescence maximum shifts from lower wavelengths to about 350 nm, the typical value for exposed tryptophan residues (Schmid, 2005).

Figure S2 compares the spectra of native and refolded  $huV_{\kappa}3$ - $huV_{H}3$  to those of the three core-grafted constructs. Spectra were recorded at 0 - 0.2 M GdmCl and at 4.9 - 5.1 M GdmCl. The emission maxima  $\lambda_{max}$  of the native and denatured state are listed in Table ST3. The decreased fluorescence quantum yield of the three interface Trp more or less balances the increased quantum yield of the conserved core Trp. As a result, the total fluorescence yield shows no drastic change upon denaturation, although the peak fluorescence intensity of the native state could not be determined with great precision. In Figure S3, the spectra of the denatured scFv are normalized to the maximal fluorescence intensity (Figure S3) and superimposed. This immediately demonstrates that not only the wavelength of the emission maximum, but also the shapes of the spectra of the denatured scFv are very similar. This finding is not compatible with the large differences in the amount of residual structure in  $huV_{\kappa}3$ - $huV_{H}3$  compared to the other constructs. However, such large differences would have

to be postulated if the denaturation curves were to be interpreted according to a two-state model (see main text).

*Equilibrium unfolding curves.* True two-state unfolding behavior of an scFv is the exception rather than the rule, particularly for scFv that require high denaturant concentrations to unfold, and it is predominantly seen in scFv that are composed of domains that have low intrinsic stability but are stabilized to a significant extent in the scFv (see below, Figure S6). We are, therefore, skeptical about the  $\Delta G_{(H2O)}$ -values derived from such extrapolations in view of the the very different m-values obtained for very similar constructs.

We thus prefer a qualitative interpretation of the unfolding curves: the unfolding curves of three of the four constructs,  $huV_{\kappa}3$ - $huV_{H}3$ ·MocB,  $huV_{\kappa}3$ - $huV_{H}3$ ·1 and  $huV_{\kappa}3$ - $huV_{H}3$ ·5, are roughly parallel in the range of their respective unfolding transitions, with m-values approaching the values expected for a two-state unfolding. The qualitative ranking of the three curves remains the same throughout the transition range. If the three spectra are compared at the same denaturant concentration, the spectral characteristics of  $huV_{\kappa}3$ - $huV_{H}3$ ·1 and  $huV_{\kappa}3$ - $huV_{H}3$ ·5 are closer to the denatured state than those of  $huV_{\kappa}3$ - $huV_{H}3$ ·MocB. The unfolding curve of the forth scFv,  $huV_{\kappa}3$ - $huV_{H}3$ , clearly shows a different slope than the other three curves — it starts unfolding at roughly the same denaturant concentration as  $huV_{\kappa}3$ - $huV_{H}3$ ·5, but reaches the fully unfolded state at a significantly higher denaturant concentration than the other three constructs. As we will argue below, this indicates a hidden intermediate.

For our constructs, this qualitative picture does not depend on the way the spectral shift has been determined (Figure S4). We compared three different methods: (i) center of spectral mass, which does not imply a specific shape of the spectrum, but is sensitive to differences in the quantum yield between the native and the unfolded state, (ii) a gaussian fit which assumes symmetry of the fluorescence peak, which may not be the case for the native state of scFv that may contain both fully buried and highly solvent accessible tryptophan residues, (iii) and fit of the spectra by a Taylor series (Monsellier and Bedouelle, 2005).

For the three chimeric constructs the ranking by  $\Delta G_{(H2O)}$ -values, determined from these curves (Tables ST1 and ST2), confirm the qualitative ranking with 11.2 kcal/mol (m = 5.1

kcal·L/mol<sup>2</sup>) for huV<sub>k</sub>3-huV<sub>H</sub>3•1, 13.0 kcal/mol (m = 5.5 kcal·L/mol<sup>2</sup>) for huV<sub>k</sub>3-huV<sub>H</sub>3•5 and 17.2 kcal/mol (m = 6.6 kcal·L/mol<sup>2</sup>) for huV<sub>k</sub>3-huV<sub>H</sub>3•MocB. While the different methods of determining the spectral fit lead to slight differences in the exact values of the thermodynamic parameters, these difference are within the experimental error and do not affect the ranking of the construct. However, a two state fit of the unfolding curve of huV<sub>k</sub>3huV<sub>H</sub>3 results in an extrapolation to a  $\Delta G_{(H2O)}$  of only 6.6 kcal/mol (m = 2.5 kcal·L/mol<sup>2</sup>) due to the very low m-value, which is not consistent with the superior functional and thermal stability observed for this scFv.

Corrective terms involving the curvature of the spectra at their respective  $\lambda_{max}$  introduced by (Monsellier and Bedouelle, 2005) to compensate for the deviation from linearity of the relation between  $\lambda_{max}$  and the fraction of unfolded molecules only account for 0.1 - 0.3 kcal/mol and have no effect on the ranking (Table ST3). These corrective turns are supposed to correct both for the different peak width of the spectrum and the different fluorescence quantum yield of the native and the denatured state. As the comparison of the normalized spectra shows (Figure S3), the difference between the shape of the spectra of the native and of the denatured state is very similar for all four constructs. In contrast, the differences in quantum yield are obscured by the variations in absolute intensity (due, e.g., to light scattering effects or concentration errors). This can be seen by the fact that the relative peak intensities differ in spectra taken from the unfolding and the refolding curve of the same construct under the same conditions (Figure S2), while the peak shape remains the same (Figure S3).

The midpoint of the equilibrium unfolding curve of the scFv fragment  $huV_{\kappa}3-huV_{H}3$  ([GdmCl]<sub>50</sub> = 2.6 M) is equal or lower than that of the isolated  $huV_{H}3$  domain ([GdmCl]<sub>50</sub> 2.7 M (Ewert *et al.*, 2003) and this study), which is typical of an scFv in which the unfolding of the two domains is not stringently coupled. The isolated  $huV_{\kappa}3$  domain unfolds with a [GdmCl]<sub>50</sub> of 2.3 M ((Ewert *et al.*, 2003) and this study). Due to aggregation of the material at low denaturant concentrations, interpretation of the unfolding curve of  $huV_{H}3$ •MocB can only be qualitative, as the fit of a pre-transition baseline was very unreliable. Comparison of the data points at higher denaturant concentrations to the unfolding curve of  $huV_{H}3$  was found to be consistent with a curve that is parallel to the unfolding curve of  $huV_{H}3$ , but shifted to 0.4 M lower GdmCl concentration. This leads to an estimate of [GdmCl]<sub>50</sub> of approximately 2.3 M for the isolated  $V_{H}$  domain. If this is correct, the midpoint for the unfolding of 2.7 M for

the scFv indicates an scFv in which both domains benefit from the association, indicating cooperative unfolding. Isolated domains  $huV_H3 \cdot 1$  and  $huV_H3 \cdot 5$  could not be produced in sufficient amounts and monomeric state to characterize of their unfolding curves.

Based on previous studies (Wörn and Plückthun, 1999; Wörn *et al.*, 2000; Jäger *et al.*, 2001; Röthlisberger *et al.*, 2005), we suggest that true two-state unfolding behavior of an scFv is the exception rather than the rule, and is predominantly seen in scFv composed of domains that have low intrinsic stability but are stabilized to a significant extent in the scFv (Figure S6a). For scFv with fairly high stabilities, unfolding curves like those depicted in Supplementary Figure S6b and S6d are more frequent: curves that show no discernable intermediate, but are characterized by a low m-value and a [GdmCl]<sub>50</sub> intermediate between those of the isolated  $V_L$  and  $V_H$  domain. If the difference between the intrinsic stabilities of the isolated and the more stable one remains folded may present itself at intermediate denaturant concentrations (Figure S6c).

The equilibrium unfolding curve of scFv fragment  $huV_{\kappa}3$ - $huV_{H}3$  is clearly not compatible with two-state equilibrium unfolding, while for the other three constructs the case is not that clear, as their m-values are closer to the expected value. This is not likely to be due to a selective destabilization of the V<sub>L</sub>/V<sub>H</sub> interface in the  $huV_{\kappa}3$ - $huV_{H}3$  construct, compared to the other three constructs: all residues whose side-chains contribute to this interface are identical in all four constructs. Furthermore, the main-chain hydrogen bonding between the strands of the  $\beta$ -sheet does not allow local changes of the backbone conformation.

In scFv that show a clear unfolding intermediate (Figure S6c), this intermediate represents a state where the variable domain with lower intrinsic stability is unfolded, while the domain with higher intrinsic stability remains folded and unfolds at the same denaturant concentration as the isolated domain (see Jäger *et al.*, 2001 for an example where the stability of  $V_H$  was limiting, Wörn and Plückthun, 1999; Wörn *et al.*, 2000 and Röthlisberger *et al.*, 2005 for a series of constructs in which  $V_L$  has been shown to be the first domain to unfold). The unfolding transition of the weaker domain is shifted to higher denaturant concentration compared to the isolated domain by the stabilizing influence of the  $V_L/V_H$  interface. Such a clear step in the unfolding trace is only seen if denaturant concentrations where the first

unfolding transition is essentially complete and the second transition has not yet started is well separated. In cases where this range is narrower and the unfolding transition of the interface-stabilized weaker domain overlaps that of the stronger domain (Figure 6b), the combined unfolding trace resembles a two-state transition with decreased m-value and a midpoint intermediate between those of the two component domains. The smaller the gap between two individual transitions, the steeper this combined unfolding trace becomes. As a result, we find that starting from an scFv like the one depicted in Figure 6b, introduction of stabilizing mutations into the weaker domain will increase the apparent  $[D]_{50}$ ,  $\Delta G_{(H2O)}$  and mvalue derived from a two-state fit. Introduction of stabilizing mutation into the stronger domain will increase the apparent [D]<sub>50</sub>, but at the same time decrease the apparent m-value, which can lead to a *decrease* of the apparent  $\Delta G_{(H2O)}$ . Introduction of *destabilizing* mutations into the more stable domain will decrease the apparent [D]<sub>50</sub>, but increases the m-value derived from a two-state fit, and therefore, paradoxically, *increases* the apparent  $\Delta G_{(H2O)}$ . The more stable an scFv is, the farther the extrapolation from  $\Delta G_{(D50)}$  to  $\Delta G_{(H2O)}$  has to reach beyond the relevant data points that contribute to the fit of the m-value, increasing the likelihood that even small changes of the m-value cancel the influence of the shift of [D]<sub>50</sub> on  $\Delta G_{(H2O)}$ .

The increased cooperativity of the scFv containing the chimeric V<sub>H</sub> domains compared to  $huV_{\kappa}3$ - $huV_{H}3$  suggests a decreased difference between the intrinsic stabilities of V<sub>L</sub> and V<sub>H</sub> in these constructs as a consequence of a reduction of the intrinsic stability of these V<sub>H</sub> domains. The decreased [D]<sub>50</sub> values for  $huV\kappa3$ - $huV_H3$ •1 and  $huV\kappa3$ - $huV_H3$ •5 compared to  $huV\kappa3$ - $huV_H3$ •MocB and  $huV\kappa3$ - $huV_H3$  point in the same direction. An alternative explanation, blaming the low cooperativity observed for  $huV_{\kappa}3$ - $huV_H3$  on a destabilization of the V<sub>L</sub>/V<sub>H</sub> interface in this particular construct is highly unlikely, since the V<sub>L</sub> domain and the V<sub>H</sub> residues contributing to the V<sub>L</sub>/V<sub>H</sub> interface are identical in all four constructs, and the main-chain hydrogen bonding between the strands of the V<sub>H</sub>  $\beta$ -sheet that supports the interface residues does not allow major changes of the backbone conformation.

We conclude that the increased cooperativity of  $huV\kappa3$ - $huV_H3$ ·MocB is a sign of a decreased intrinsic stability of  $huV_H3$ ·MocB compared to  $huV_H3$ . Therefore, in the context of the  $huV_H3$  CDRs, the  $huV_H3$  consensus framework is more stable than the  $huV_H3$ ·MocB chimera, while in the context of the Moc31-derived CDRs, the  $huV_H3$ ·MocB chimeric framework was more

stable than the consensus framework. This difference in the stability ranking between the two frameworks can be explained by a different amount of steric strain between framework and CDRs.

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Figure S1: Positions of tryptophan residues in the scFv fragment: Constructs  $huV_{\kappa}^3$ - $huV_H^3$ ,  $huV_{\kappa}^3$ - $huV_H^3$ •MocB and  $huV_{\kappa}^3$ - $huV_H^3$ •1 contain 5 Trp residues: Trp L43 is fully quenched in the native state, the same is probably true for Trp H43. Three Trp residues are mostly buried in the interface between  $V_L$  and  $V_H$ : Trp H54 (conserved), Trp H109 (CDR-H3) and Trp H139 (conserved). They become solvent exposed when either domain unfolds. Construct  $huV_{\kappa}^3$ - $huV_H^3$ •5 contains a sixth Trp, H93, which is fully buried in the  $V_H$  core.



**Figure S2:** Fluorescence spectra of the scFv fragments  $huV_{\kappa}3$ - $huV_{H}3$  (a),  $huV_{\kappa}3$ - $huV_{H}3$ •MocB (b) and  $huV_{\kappa}3$ - $huV_{H}3$ •1 (c) and  $huV_{\kappa}3$ - $huV_{H}3$ •5 (d) at low (0 - 0.2 M GdmCl (n)) and high (4.9-5.1 M GdmCl, (u)) denaturant concentration. Comparison of the corresponding spectra taken from the unfolding curve (e.g.  $huV\kappa3$ - $huV_{H}3u$ ) and from the refolding curve (e.g.  $huV\kappa3$ - $huV_{H}3r$ ) demonstrate the variations of the fluorescence intensities, but the reproducibility and reversibility of  $\lambda_{max}$  (cf. the normalized spectra in Figure S3)



**Figure S3: Normalized fluorescence spectra.** The spectra shown in Figure S2 were normalized at the maximal fluorescence intensity to show that the shape and wavelength of the fluorescence maxima of the unfolded spectra is the same for all constructs, while the wavelength of the fluorescence maxima of the native spectra differ slightly for the different constructs.



**Figure S4: Equilibrium unfolding curves.** For each of the four scFv in the study, both the equilibrium unfolding curves (u) and the equilibrium refolding curves (r) starting from material equilibrated for 24 h in 6 M GdmCl were determined. The spectral shifts upon denaturation were determined by different methods and fitted according to eq. 1 (main text, Materials and Methods) (Creighton, 1997). The parameters derived from this fit are listed in Table ST1

(a) Center of spectral mass (csm) plotted against the denaturant concentration. The csm over the interval from 320 to 370 nm is determined as the sum of each fluorescence intensity at a given wavelength multiplied by this wavelength, divided by the sum of the fluorescence intensities over this interval. The csm makes no assumptions concerning the shape of the fluorescence spectrum.

(b) Wavelength of the emission maximum determined by fitting a Gaussian curve to the fluorescence peak over the interval from 320 to 370 nm. This is a deliberately poor choice, as we usually use a narrower interval to more precisely fit the peak maximum, but it still leads to very similar curves.

(c) Wavelength of the emission maximum determined by fitting a Taylor series to the third order term (Monsellier and Bedouelle, 2005)

(d) Wavelength of the emission maximum determined by fitting a Taylor series to the fourth order term (Monsellier and Bedouelle, 2005)



Figure S5: Curvature of the spectra shown in Figures S2 and S3.  $\lambda_{max}(n)$  and  $\lambda_{max}(u)$  of each construct were determined by fitting a Taylor series to the fourth order term to the spectra in the wavelength interval from 320 to 370 nm. The curvature was then determined by fitting the spectra with a parabola, centered on  $\lambda_{max}(n)$  or  $\lambda_{max}(u)$ , respectively, in the wavelength interval from  $\lambda_{max}(n) - 2$  nm to  $\lambda_{max}(u) + 2$  nm according to Monsellier and Bedouelle (2005). (a) Spectra of native scFv, (b) spectra of denatured scFv (c) spectra of native scFv normalized to 100% intensity at  $\lambda_{max}$ , (d) spectra of denatured scFv normalized to 100% intensity at  $\lambda_{max}$ . The parameters derived from (a) and (b) are listed in Table ST3, together with the  $\Delta G_{(H2O)}$ -correction derived from this fit according to Monsellier and Bedouelle (2005). The values derived from (c) and (d) are given in Table ST3 for illustration only.



**Figure S6: Two-state and non two-state systems.** Cooperativity of scFv unfolding curves as a function of the intrinsic stabilities of the constituent domains (redrawn from the data of Wörn and Plückthun (1999); Röthlisberger *et al.*, (2005)).

(a) Unfolding of a weak  $V_L$  domain (*w.t.* AB48) is coupled to a weak  $V_H$  domain (disulfidefree AB48 Cys H23 Ala) in  $V_H$ -linker- $V_L$  orientation. The resulting scFv shows a cooperative equilibrium unfolding curve with a midpoint at significantly higher denaturant concentration than the midpoints of either of the two constituent domains (indicated by dashed lines).

(b) A weak  $V_L$  domain (*w.t.* AB48) is coupled to a  $V_H$  domain of intermediate stability (disulfide-free AB48 Lys H77 Arg, Asn H59 Ser, Tyr H106 Val). The resulting scFv shows an equilibrium unfolding curve with apparent low cooperativity (broad transition) and a midpoint denaturant concentration intermediate between those of the two constituent domains.

(c) A weak  $V_L$  domain is coupled to a very strong  $V_H$  domain (disulfide-restored AB48 Lys H77 Arg, Asn H59 Ser). The unfolding curve of the scFv shows a clear equilibrium unfolding intermediate; although the unfolding curve of the  $V_L$  domain is shifted to higher denaturant concentrations than that of the free  $V_L$  domain, the  $V_L$  in the scFv fully unfolds at denaturant

concentrations at which  $V_H$  has not even started to unfold. The unfolding curve of the  $V_H$  domain within the scFv is not affected by the presence of the  $V_L$  domain.

(d) A very strong  $V_L$  domain (4D5) is coupled to a very strong  $V_H$  domain (disulfide-restored AB48 Lys H77 Arg, Asn H59 Ser). The resulting scFv shows an equilibrium unfolding curve with apparent low cooperativity and a midpoint denaturant concentration intermediate between those of the two constituent domains



Figure S7: Fluorescence emission spectra of the isolated domains  $huV\kappa 3$  and  $huV_H3$  in native and denatured form, compared to the scFv huV $\kappa$ 3-huV<sub>H</sub>3. The scales of the three panels have been adjusted to make the peak intensity of the fully denatured proteins (red curves, (u)), proportional to the number of tryptophan residues in each construct (since those spectra are essentially dominated by exposed Trp). The spectra of the native proteins (blue curves, (n)) are properly scaled in relation to their denatured counterparts, as experimentally determined. (a) The  $V_L$  domain contains a single Trp (L43) that is fully quenched in the native state. (b) The fluorescence spectrum of the native  $V_H$  domain (4 Trp) is dominated by the three Trp that in the scFv are buried in the  $V_L/V_H$  interface. In the isolated domain, these Trp are partially solvent-exposed. The core Trp H43 presumably is quenched to a similar extent as the core Trp of the  $V_L$  domain. (c) The intensity of the fluorescence emission of the native scFv is significantly higher than the sum of the emissions of the isolated domains. Since the three interface Trp are mostly buried in the scFv, the wavelength of the emission maximum of the native scFv is shifted to lower wavelengths compared to the isolated V<sub>H</sub> domain. Upon unfolding of the scFv fragment, two effects come into play: first, the core Trp residues of V<sub>L</sub> and V<sub>H</sub> become unquenched, resulting in an increased fluorescence of the isolated domains. Second, fluorescence quantum yield of the interface Trp residues decreases, when this interface is destroyed. Both effects somewhat compensate each other, such that the overall quantum yield at the respective maxima of the native and unfolded scFv fragment is not very different (blue and red curves in (c)). See Figure S1 for the location of the Trp residues in the scFv structure. Fitted  $\lambda$ max and curvatures are listed in Table ST4. The Figure was produced using raw data from Ewert et al. (2003).

## Table ST1: Fit of a two-state model of the unfolding transition to the data depicted in Figure S4, using different fitting approaches

		$\lambda \max(n)$	slope(n)	λ max(u)	slope (u)	∆G(H2O)	m	R
		nm	nm*L/mol	nm	nm*L/mol	J/mol	J*L/mol^2	
а	huVκ3-huVH3u	343.1 ± 0.03	-0.078 ± 0.029	345.9 ± 0.17	0.053 ± 0.039	27610 ± 1139	10394 ± 430	0.9993
	huVx3-huVH3r	343.2 ± 0.03	-0.127 ± 0.026	345.9 ± 0.12	0.038 ± 0.026	25597 ± 820	9688 ± 309	0.9996
	huVκ3-huVH3●MocBu	342.8 ± 0.04	0.058 ± 0.028	345.7 ± 0.17	0.083 ± 0.043	67230 ± 6181	25608 ± 2368	0.9989
	huVκ3-huVH3●MocBr	342.9 ± 0.05	0.055 ± 0.036	345.4 ± 0.10	0.147 ± 0.025	69012 ± 5019	26384 ± 1898	0.9990
	huVĸ3-huVH3●1u	343.3 ± 0.03	-0.029 ± 0.027	345.5 ± 0.07	0.130 ± 0.019	45284 ± 1998	20528 ± 901	0.9995
	huVĸ3-huVH3●1r	343.3 ± 0.06	0.020 ± 0.063	345.5 ± 0.14	0.113 ± 0.034	44677 ± 4952	20214 ± 2212	0.9976
	huVκ3-huVH3●5u	343.2 ± 0.03	-0.050 ± 0.023	345.8 ± 0.05	0.052 ± 0.012	51072 ± 2034	21556 ± 850	0.9995
	huVκ3-huVH3●5r	343.2 ± 0.04	-0.004 ± 0.030	345.7 ± 0.04	0.078 ± 0.010	54612 ± 2604	23254 ± 1097	0.9993
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D	huVκ3-huVH3u	338.7 ± 0.13	-0.301 ± 0.123	348.6 ± 0.68	0.370 ± 0.159	29328 ± 1376	11203 ± 527	0.9991
	huVκ3-huVH3r	339.2 ± 0.13	-0.430 ± 0.125	348.9 ± 0.51	0.289 ± 0.118	27758 ± 1159	10627 ± 440	0.9993
	huVκ3-huVH3●MocBu	337.5 ± 0.12	0.314 ± 0.073	348.7 ± 0.45	0.369 ± 0.113	79782 ± 5718	30447 ± 2185	0.9994
	huVκ3-huVH3●MocBr	337.8 ± 0.15	0.182 ± 0.112	347.9 ± 0.30	0.533 ± 0.077	70040 ± 4038	26903 ± 1536	0.9994
	huVĸ3-huVH3●1u	339.5 ± 0.11	-0.093 ± 0.113	348.1 ± 0.30	0.525 ± 0.078	48651 ± 2437	22191 ± 1107	0.9994
	huVĸ3-huVH3●1r	339.3 ± 0.28	0.102 ± 0.272	348.3 ± 0.56	0.436 ± 0.142	45981 ± 5648	20965 ± 2537	0.9972
	huVκ3-huVH3●5u	339.2 ± 0.10	-0.225 ± 0.088	349.0 ± 0.19	0.223 ± 0.045	53020 ± 2123	22510 ± 892	0.9995
	huVκ3-huVH3●5r	338.8 ± 0.12	0.064 ± 0.106	348.6 ± 0.15	0.301 ± 0.034	55022 ± 2438	23587 ± 1033	0.9994
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С	huVκ3-huVH3u	337.5 ± 0.14	-0.317 ± 0.128	349.0 ± 0.76	0.399 ± 0.177	29282 ± 1287	11016 ± 486	0.9992
	huVx3-huVH3r	337.8 ± 0.16	-0.351 ± 0.139	349.6 ± 0.65	0.221 ± 0.149	28695 ± 1236	10742 ± 461	0.9993
	huVκ3-huVH3●MocBu	336.3 ± 0.13	0.266 ± 0.080	349.0 ± 0.50	0.389 ± 0.123	77096 ± 5218	29323 ± 1986	0.9995
	huVκ3-huVH3●MocBr	336.5 ± 0.24	0.192 ± 0.177	347.7 ± 0.48	0.670 ± 0.123	72313 ± 6046	27641 ± 2288	0.9987
	huVĸ3-huVH3●1u	338.3 ± 0.13	-0.067 ± 0.124	348.3 ± 0.34	0.590 ± 0.088	49417 ± 2353	22404 ± 1062	0.9994
	huVĸ3-huVH3●1r	338.1 ± 0.29	0.148 ± 0.275	348.4 ± 0.58	0.537 ± 0.148	47654 ± 5206	21600 ± 2331	0.9978
	huVκ3-huVH3●5u	337.9 ± 0.12	-0.196 ± 0.099	348.9 ± 0.23	0.296 ± 0.053	54483 ± 2272	22996 ± 951	0.9995
	huVκ3-huVH3●5r	337.6 ± 0.13	0.067 ± 0.111	348.5 ± 0.15	0.389 ± 0.036	58312 ± 2390	24818 ± 1007	0.9995
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a	huVκ3-huVH3u	338.4 ± 0.13	-0.267 ± 0.118	347.1 ± 0.65	0.493 ± 0.152	29041 ± 1427	11127 ± 549	0.9990
	huVx3-huVH3r	338.7 ± 0.15	-0.155 ± 0.139	348.0 ± 0.61	0.232 ± 0.140	28832 ± 1556	10895 ± 585	0.9989
	huVκ3-huVH3∙MocBu	337.2 ± 0.13	0.301 ± 0.084	347.6 ± 0.52	0.379 ± 0.130	75301 ± 6173	28779 ± 2369	0.9992
	huVκ3-huVH3∙MocBr	337.4 ± 0.18	0.203 ± 0.133	347.0 ± 0.35	0.492 ± 0.091	68696 ± 4876	26417 ± 1857	0.9990
	huVĸ3-huVH3●1u	339.2 ± 0.12	-0.056 ± 0.120	347.3 ± 0.32	0.468 ± 0.083	47623 ± 2676	21729 ± 1216	0.9992
	huVx3-huVH3●1r	338.9 ± 0.28	0.220 ± 0.268	347.4 ± 0.55	0.425 ± 0.140	46599 ± 6188	21242 ± 2780	0.9969
	huVx3-huVH3●5u	338.8 ± 0.09	-0.214 ± 0.078	347.9 ± 0.17	0.223 ± 0.039	50386 ± 1838	21490 ± 775	0.9996
	huVx3-huVH3●5r	338.8 ± 0.13	-0.123 ± 0.104	347.6 ± 0.12	0.286 ± 0.027	54184 ± 2036	23243 ± 861	0.9996

\*u = unfolding \*r = refolding from 6M GdmCl

The columns give the fitted parameters form a two-state interpretation of the data for the pretransition baseline (intercept  $\lambda_{max}(n)$  and slope(n)), for the post-transition baseline (intercept  $\lambda_{max}(u)$  and slope(u)), the apparent free energy of unfolding  $\Delta G_{(H2O)}$  extrapolated to zero denaturant and the apparent m-value (slope of extrapolation). Eq. 1 was used (in the main text), using the change of the relevant fluorescence wavelength as a function of denaturant. The letters (a, b, c, d) indicate which wavelength was used:

- (a) Center of spectral mass of the fluorescence spectrum from 320 nm 370 nm as a function of denaturant concentration
- (b) Wavelength of the fluorescence maximum determined by a Gaussian fit of the fluorescence spectrum from 320 nm 370 nm as a function of denaturant concentration
- (c) Wavelength of the fluorescence maximum determined from a fit of the fluorescence spectrum from 320 nm 370 nm by a Taylor expansion to the 3<sup>rd</sup> power term as a function of denaturant concentration
- (d) Wavelength of the fluorescence maximum determined from a fit of the fluorescence spectrum from 320 nm 370 nm by a Taylor expansion to the 4<sup>th</sup> power term as a function of denaturant concentration.

Table ST2: Comparison of the apparent  $\Delta G_{(H2O)}$ , m-values and  $[GdmCl]_{50}$  derived from the fits shown in Figure S4 and Table ST1

		а	b	С	d
∆ <b>G(H2O)</b>	huVx3-huVH3u	6.6 ± 0.27	7.0 ± 0.33	<b>7.0</b> ± 0.31	6.9 ± 0.34
kcal/mol	huVκ3-huVH3r	6.1 ± 0.20	6.6 ± 0.28	<b>6.9</b> ± 0.30	6.9 ± 0.37
	huVκ3-huVH3●MocBu	<b>16.1</b> ± 1.48	19.1 ± 1.37	18.4 ± 1.25	18.0 ± 1.47
	huVκ3-huVH3●MocBr	<b>16.5</b> ± 1.20	16.7 ± 0.96	17.3 ± 1.44	16.4 ± 1.16
	huVĸ3-huVH3●1u	<b>10.8</b> ± 0.48	<b>11.6</b> ± 0.58	<b>11.8</b> ± 0.56	11.4 ± 0.64
	huVĸ3-huVH3●1r	<b>10.7</b> ± 1.18	<b>11.0</b> ± 1.35	<b>11.4</b> ± 1.24	<b>11.1</b> ± 1.48
	huVĸ3-huVH3●5u	<b>12.2</b> ± 0.49	<b>12.7</b> ± 0.51	<b>13.0</b> ± 0.54	12.0 ± 0.44
	huVĸ3-huVH3●5r	<b>13.0</b> ± 0.62	<b>13.1</b> ± 0.58	<b>13.9</b> ± 0.57	<b>12.9</b> ± 0.49
m	huVĸ3-huVH3u	<b>2.5</b> ± 0.10	<b>2.7</b> ± 0.13	<b>2.6</b> ± 0.12	<b>2.7</b> ± 0.13
kcal*L/mol^2	huVĸ3-huVH3r	<b>2.3</b> ± 0.07	<b>2.5</b> ± 0.11	<b>2.6</b> ± 0.11	<b>2.6</b> ± 0.14
	huVĸ3-huVH3●MocBu	<b>6.1</b> ± 0.57	<b>7.3</b> ± 0.52	<b>7.0</b> ± 0.47	6.9 ± 0.57
	huVκ3-huVH3●MocBr	6.3 ± 0.45	6.4 ± 0.37	6.6 ± 0.55	6.3 ± 0.44
	huVĸ3-huVH3●1u	<b>4.9</b> ± 0.22	5.3 ± 0.26	5.4 ± 0.25	5.2 ± 0.29
	huVĸ3-huVH3●1r	<b>4.8</b> ± 0.53	<b>5.0</b> ± 0.61	<b>5.2</b> ± 0.56	<b>5.1</b> ± 0.66
	huVĸ3-huVH3●5u	<b>5.1</b> ± 0.20	<b>5.4</b> ± 0.21	<b>5.5</b> ± 0.23	<b>5.1</b> ± 0.19
	huVĸ3-huVH3●5r	<b>5.6</b> ± 0.26	5.6 ± 0.25	<b>5.9</b> ± 0.24	<b>5.6</b> ± 0.21
[D]50	huVĸ3-huVH3u	<b>2.7</b> ± 0.17	<b>2.6</b> ± 0.20	<b>2.7</b> ± 0.19	<b>2.6</b> ± 0.21
mol/L	huVĸ3-huVH3r	<b>2.6</b> ± 0.12	<b>2.6</b> ± 0.17	<b>2.7</b> ± 0.18	<b>2.6</b> ± 0.23
	huVĸ3-huVH3●MocBu	<b>2.6</b> ± 0.91	<b>2.6</b> ± 0.84	<b>2.6</b> ± 0.77	<b>2.6</b> ± 0.91
	huVκ3-huVH3●MocBr	<b>2.6</b> ± 0.75	<b>2.6</b> ± 0.60	<b>2.6</b> ± 0.90	<b>2.6</b> ± 0.72
	huVĸ3-huVH3●1u	<b>2.2</b> ± 0.26	<b>2.2</b> ± 0.32	<b>2.2</b> ± 0.31	<b>2.2</b> ± 0.35
	huVĸ3-huVH3●1r	<b>2.2</b> ± 0.65	<b>2.2</b> ± 0.74	<b>2.2</b> ± 0.69	<b>2.2</b> ± 0.81
	huVĸ3-huVH3●5u	<b>2.4</b> ± 0.28	<b>2.4</b> ± 0.29	<b>2.4</b> ± 0.32	<b>2.3</b> ± 0.25
	huVx3-huVH3●5r	<b>2.3</b> ± 0.36	<b>2.3</b> ± 0.34	<b>2.3</b> ± 0.33	<b>2.3</b> ± 0.28

\*u = unfolding

\*r = refolding from 6M GdmCl

The letters a, b, c and d indicate the method used to fit the spectra, as described in the legends to Figure S4 and Table ST1

Table ST3: Corrections to  $\Delta G_{(H2O)}$  -values derived from the spectral shift for the different curvatures of the spectra of the native and the denatured scFv according to Monsellier and Bedouelle (2005).

						RT*ln(b(	n)/(bu))
	λmax (n)	λmax (u)	b(n)	b(u)	b(n)/b(u)	J/mol	kcal/mol
huVκ3-huVH3u	338.3	349.3	-1059	-1109	0.955	-113	-0.03
huVx3-huVH3r	338.6	349.0	-949	-1118	0.849	-400	-0.10
huVκ3-huVH3●MocBu	337.3	349.3	-1085	-1174	0.924	-192	-0.05
huVκ3-huVH3●MocBr	337.5	349.2	-814	-1138	0.715	-816	-0.19
huVκ3-huVH3●1u	339.2	349.5	-1328	-1434	0.926	-188	-0.04
huVκ3-huVH3●1r	339.1	349.2	-979	-1183	0.827	-462	-0.11
huVκ3-huVH3●5u	338.7	349.0	-1061	-1412	0.751	-697	-0.17
huV <b>ĸ3-huVH3</b> ●5r	338.6	348.9	-728	-1219	0.597	-1258	-0.30

#### Curvatures derived from raw spectra

Curvatures derived from normalized spectra (for illustration only)

						RT*in(i	on/bu)
	λmax (n)	λ <b>max (u)</b>	b(n)	b(u)	b(n)/b(u)	J/mol	kcal/mol
huVκ3-huVH3u	338.3	349.3	-0.135	-0.163	0.831	-450	-0.11
huVx3-huVH3r	338.6	349.0	-0.136	-0.165	0.825	-469	-0.11
huVκ3-huVH3●MocBu	337.3	349.3	-0.130	-0.161	0.806	-526	-0.13
huVκ3-huVH3●MocBr	337.3	349.2	-0.133	-0.163	0.815	-498	-0.12
huVκ3-huVH3●1u	337.5	349.5	-0.132	-0.160	0.821	-480	-0.11
huVκ3-huVH3●1r	339.1	349.2	-0.129	-0.161	0.802	-537	-0.13
huVκ3-huVH3●5u	338.7	349.0	-0.130	-0.164	0.796	-555	-0.13
huVκ3-huVH3●5r	338.6	348.9	-0.129	-0.164	0.789	-578	-0.14

\*u =unfolding

\*r = refolding from 6M GdmCl

The terms b(n) and b(u) are defined in Monsellier and Bedouelle (2005). Curves derived from normalized spectra are only shown to demonstrate that the variability in the curvature-derived  $\Delta G$  correction terms is entirely due to the large error in the fluorescence intensities and not due to changes in the shape of the spectrum.

Table ST4: Emission maxima of the native and the unfolded state and curvatures for the native and denatured isolated  $V_L$  and  $V_H$  domain shown in Figure S7 compared to the scFv.

	λ <b>max(n)</b>	λ <b>max(u)</b>	b(n)	b(u)	b(n)/b(u)
huVk3	n.d.	348.3	n.d.	-0.121	n.d.
huVH3	345.1	350.4	-0.585	-0.667	0.877
huVk3-huVH3	338.5	347.2	-0.747	-0.814	0.918

Since the core Trp is fully quenched in the native state of huV $\kappa$ 3 (see Figure S7),  $\lambda$ max (n), b(n) and b(n)/b(u) could not be determined (nd). Therefore, the spectral shift cannot be used to evaluate the stability of the isolated V<sub>L</sub> domain. However, isolated V<sub>L</sub> domains are much less prone to aggregation than V<sub>H</sub> domains, permitting the use of fluorescence intensity to monitor unfolding. Since the interface Trp are solvent exposed in the isolated V<sub>H</sub> domain, the spectral shift upon unfolding is much smaller than for the scFv.