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# An Escherichia coli Protein Consisting of a Domain Homologous to **FK506-binding Proteins (FKBP) and a New Metal Binding Motif\***

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Initially detected as a persistent contaminant in immobilized metal affinity chromatography of recombinant proteins in *Escherichia coli*, a 196-amino acid protein was isolated, cloned, overexpressed, and characterized. It consists of two domains, of which the first (146 amino acids) shows some homology to the FK506binding proteins. The second domain (50 amino acids) is extremely rich in potentially metal-binding amino acids, such as histidine, cysteine, and acidic amino acids. The protein binds  $Ni^{2+}$  and  $Zn^{2+}$  tightly with 1:1 stoichiometry, Cu<sup>2+</sup> and Co<sup>2+</sup> with lower affinity, and Mn<sup>2+</sup>, Fe<sup>2+</sup>, Fe<sup>3+</sup>, Mg<sup>2+</sup>, and Ca<sup>2+</sup> hardly at all.

ing proteins (Harding et al., 1986; Siekierka et al., 1989) was detected, a class of proteins involved in immunosuppression. Two classes of such proteins have been described, the other being cyclophilins (Fischer et al., 1984) which bind the immunosuppressant cyclosporin A. The two classes are related neither by sequence homology nor folding topology, nor is there any conspicuous structural homology between the immunosuppressants FK506 and cyclosporin A. Both classes do show proline *cis-trans*-isomerase (rotamase) activity and can accelerate the folding of some proteins in vitro. The biological significance of this *in vitro* reaction for immunosuppression remains a mystery, however. Rather, the complex of immunosuppressant and immunophilins seems to react with phosphatases (Liu et al., 1991). Both classes include bacterial homologs (Hayano et al., 1991; Liu and Walsh, 1990; Perry et al., 1988) that are not all able to bind the respective immunosuppressants.

Metals can be bound by a large number of diverse proteins. They may have an exclusively structural role as e.g. in the zinc-finger domains of several transcription factors, or they may be involved in the catalytic function of the protein. Most metals can serve structural functions, but redox-active metals are less likely to do so, since the change in redox state can change the preferences for coordination geometry. Catalytic functions of metals include redox and Lewis acid chemistry, both of which are more likely to be carried out by transition group metals due to their ease of change of redox state and their polarizing capability.

During extensive work on the purification of several heterologous recombinant proteins from E. coli by immobilized metal affinity chromatography (IMAC)<sup>1</sup> (Porath, 1987; Arnold, 1991; Lindner et al., 1992), a host-encoded protein was observed as the major contaminant and could be isolated and purified to homogeneity. In view of its metal binding ability expected from its binding to the IMAC column, and taking into account the general importance of transition metal binding proteins in E. coli (Beveridge, 1989), this protein was cloned. We found that it consists of two domains, an NH<sub>2</sub>-terminal domain of about 150 amino acids and a COOH-terminal metal binding motif of about 50 amino acids.

We describe here the isolation, cloning, sequence, overexpression, and native purification of the protein. It was characterized by CD spectroscopy, and its metal binding properties were investigated.

#### MATERIALS AND METHODS

DNA Manipulations—DNA manipulations were performed according to standard protocols (Sambrook et al., 1989). The PCR oligonucleotides derived from the peptide sequences were ACT GAG AAT TCT AYC ARG TN(C/A) GNA CNG ARG A and CGG ACG TCG ACY TCY TCY TCN GTN GCY TC. The product of the first PCR with the degenerate oligonucleotides and XbaI-cleaved genomic DNA was gel-purified and cloned into the sequencing vector pSL301 (Invitrogen Corp., San Diego, CA) after cleavage with the restriction endonucleases EcoRI and XbaI. Inverse PCR (Silver, 1991) was performed in two parts after determining suitable restriction enzymes by Southern blotting (Gebeyehu et al., 1987). Using the restriction endonuclease AflIII, which cuts within the already known part of the gene for the preparation of the inverse PCR template, two independent fragments, a 800-base pair fragment for the NH<sub>2</sub> terminus and a 3.5-kilobase pair fragment for the COOH terminus, were amplified. The PCR oligonucleotides for the inverse PCR amplification of the NH<sub>2</sub> terminus were ACT GAG AAT TCG GCG CGA ACG ACG CTT and ACT GAT CTA GAT CAC CGG AGA CTC ATC AA, and the ones for the COOH terminus were ACT GAG AAT TCT GAA ATT CAA CGT TGA AGT and ACT GAT CTA GAC CCA TAA ATA CGT CTT TAG G. The inverse PCR for the NH<sub>2</sub> terminus was performed following Ponce and Micol (1992), and the reaction product was cloned as above. For the COOH terminus the 3.5-kilobase pair DNA fragment was amplified following Kainz et al. (1992). After cleavage with NruI and NsiI and fill-in of the NsiI ends, the reaction product was cloned blunt-ended into pSL301. In order to confirm the WHP sequence, the gene was PCR-amplified three times directly from genomic DNA using the inward oligonucleotides ACT GAT CTA GAG GCT GAA GAAACG CCA and ACT GAG AAT TCC GCT ACA ATC TGC G. After cloning via EcoRI and XbaI in pSL301, the three PCR products were sequenced. In order to express WHP with sufficient yield, the gene was ligated into the SphIATG-cloning site of pUHE25-2.<sup>2</sup> For this purpose, the gene of WHP was cut out of the sequencing vector with EcoRV and XbaI and blunt-ended. The expression vector was termed pWHPexpr.

During the characterization of this protein a weak, but significant homology of the NH2-terminal domain to FK506-bind-

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The nucleotide sequence(s) reported in this paper has been submitted to the GenBank<sup>TM</sup>/EMBL Data Bank with accession number(s) Z21496. <sup>‡</sup> Present address: Center of Immunology, P. O. Box 16040, Havana 11600, Cuba.

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<sup>1</sup> The abbreviations used are: IMAC, immobilized metal affinity chromatography; PCR, polymerase chain reaction; iPCR, inverse PCR; MOPS, 4-morpholinepropanesulfonic acid; PAGE, polyacrylamide gel electrophoresis; AAS, atomic absorption spectroscopy; ORF, open reading frame; FKBP, FK506-binding proteins.

<sup>2</sup> H. Bujard, personal communication.



Protein Purification—A 2-liter culture of E. coli strain RB791 (Brent and Ptashne, 1981) harboring plasmid pWHPexpr was grown to  $OD_{550}$ = 0.5, the expression of WHP was induced with 1 mm isopropyl-1-thio- $\beta$ -D-galactopyranoside, and the cells were grown for another 3 h at 37 °C. After cell rupture in a French pressure cell, the soluble fraction was loaded onto a Ni<sup>2+</sup>-NTA column (25 ml), equilibrated with 100 mm Tris, pH 8.0, 1 м NaCl. After washing the column with 500 ml of equilibration buffer containing 8 mm imidazole, WHP was eluted with a 500-ml gradient from 8 to 100 mM imidazole. The column was run at a flow rate of 60 ml/h. The pooled WHP fractions were concentrated by ultrafiltration, and the buffer was changed on a PD10 column (Pharmacia) to 10 mm MOPS, pH 7.0. The sample was then loaded onto a DEAE-Sepharose column (66 ml) (Pharmacia), washed with 100 ml of 10 mm MOPS, pH 7.0, and the column was run with a gradient from 0 to 1 м NaCl. WHP eluted at about 0.5 м NaCl. The column was run at a flow rate of 35 ml/h.

Alternatively, a hydroxylapatite column was used as the first chromatography step. 500 ml of E. coli cells were grown as above. After centrifugation the bacterial cell pellet was resuspended in 1 mм NaCl. After cell rupture as above, the soluble fraction was loaded onto a 125-ml hydroxylapatite column (Bio-Rad). The column was washed, first with 200 ml of 1 M NaCl, then with 200 ml of 10 mM sodium phosphate, pH 6.8. WHP was eluted with a gradient of 600 ml sodium phosphate (10-300 mм), pH 6.8. As a second purification step a DEAE-Sepharose column was used as above. Rotamase Activity Assay—The assay was performed according to Liu et al. (1990). WHP purified by IMAC was assayed at a final protein concentration of 0.2 µM without a detectable rate acceleration in the isomerization reaction. As a positive control, E. coli periplasmic peptidyl proline *cis-trans*-isomerase gave a strong acceleration in the assay already at 50-fold lower molar concentrations. Computer Methods-Data base homology searches were performed using the UWGCG (University of Wisconsin Genetics Computer Group) and PIR (Protein Identification Resource) program packages. The FASTA algorithm (Pearson and Lipman, 1988) was used for simple sequence homology searches and PROFILESEARCH (Gribskov et al., 1989) to search for homology to a sequence profile, generated with PILEUP. This same algorithm was used to find sequences compatible with the three-dimensional profile, determined for the FK506-binding protein structure (van Duyne et al., 1991) by the method of Eisenberg and co-workers (Lüthy et al., 1992). The CLUSTAL (Higgins et al., 1992) program was used for clustered sequence alignment. The isoelectric point was calculated using the UWGCG program package.

ment disrupts the structure of essentially all proteins, an affinity to the column would require  $Ni^{2+}$ -binding residues clustered not only in the structure, but also in the sequence. This fact seemed to be unusual enough to investigate the protein further.

The protein was subjected to  $NH_2$ -terminal protein sequencing and amino acid analysis. A peptide from a tryptic digest was also isolated using an anhydro-trypsin column (Ishii *et al.*, 1983), which usually results in the isolation of the COOHterminal peptide, and it was sequenced as well.

For the PCR cloning of the gene of WHP from genomic DNA, two sets of degenerate oligonucleotides were used. The first was derived from the amino acid sequence of the  $NH_2$  terminus of WHP, the second from the sequence of the major tryptic peptide in the run-through of the anhydro-trypsin affinity column (Ishii et al., 1983), assuming that this peptide was the COOH-terminal one. A PCR fragment of only 450 base pairs was obtained, cloned, and sequenced, but its length indicated that the complete WHP could not have been encoded on this fragment, and a considerable part of the COOH terminus of the gene was still missing. Southern blotting using the cloned PCR fragment as a probe was then carried out to determine which restriction enzymes give rise to fragments suitable for inverse PCR (iPCR) (Silver, 1991) which might contain the whole gene of WHP. Attempts to clone fragments from iPCR longer than 1200 base pairs failed. Only clones with large deletions could be isolated, indicating that a gene upstream or downstream of the WHP gene is sensitive to overexpression in the homologous system. To avoid this problem, the gene was assembled from parts using inverse PCR of restriction digests of genomic DNA that placed the respective termini of the WHP gene on different restriction fragments. The NH<sub>2</sub>-terminal part of the gene could be cloned without problems; for the COOH terminus the iPCR product had to be shortened by restriction digestion to permit cloning. The gene sensitive to overexpression is thus located downstream of the WHP gene.

CD Spectroscopy—Samples were dialyzed against 5 mm sodium phosphate buffer, pH 7.5, before recording the spectrum. The concentration of WHP was 0.83 mg/ml. The spectrum was recorded between 192 and 250 nm. An Auto Dicrograph Mark IV Spectrometer from Jobin Yvon Division Instruments SA was used. The data were processed using the OMA software from Instruments SA (Provencher, 1982), including the CONTIN program to calculate secondary structure contents. Metal Binding Studies—Protein concentrations were determined using OD<sub>280</sub> with an extinction coefficient of  $\epsilon = 5.6$  (mm cm)<sup>-1</sup> (Gill and von Hippel, 1989). For each metal binding assay, 1 mg of native IMACpurified protein was dialyzed overnight against 1 mm EDTA, 20 mm Tris, 10 mm citrate, 65 mm NaCl, 250 mm KCl, pH 7.6. Subsequently, metal ions were added at a final concentration of 2 mm, and the metal was allowed to bind overnight. Excess metal was removed by gel filtration on a Sephadex G-25 column (50 cm  $\times$  1.4 cm). The column was equilibrated, washed, and eluted with 300 mm NaCl, 20 mm Tris, pH 7.5. The Fe<sup>2+</sup> samples were allowed to equilibrate under rigorous exclusion of air in the presence of 20 mm ascorbate to prevent oxidation to Fe<sup>3+</sup>. In one  $Fe^{2+}$  sample phenanthroline was present at a concentration of 10 mm to prevent precipitation of iron hydroxides. Finally, the metal content after removal of unbound metal ions by gel filtration was determined by atomic absorption spectroscopy on a 1100B Spectrometer from Perkin Elmer. As controls, samples taken before and after incubation with EDTA were measured as well. Absence of proteolytic degradation was confirmed by running an aliquot of the AAS samples on an SDS-PAGE gel.

The sequence information obtained by iPCR was used to amplify and clone the whole WHP gene directly from the E. coli genome using inward primers. The final sequence of the coding region was determined from three independent inward PCRs and the inverse PCR reaction. The NH<sub>2</sub>-terminal part of the gene includes a putative translation initiation region and a possible promoter sequence (Fig. 1) with significant homology to E. coli  $\sigma^{70}$  promoters. The amino acid composition derived from translation of the gene corresponds to the experimentally determined one (data not shown), and the sequences of the two peptides, determined by Edman degradation (shown *underlined* in Fig. 1), were found in the DNA sequence. Comparing DNA and protein sequences, WHP does not have a signal peptide, it thus is a cytoplasmic protein. The existence of a transcription terminator consensus sequence right behind the stop codon (Fig. 1) shows that the complete gene has been cloned and that the gene is a separate transcription unit. An apparent molecular mass of 26 kDa on SDS-PAGE corresponds well with an acidic 196-amino acid protein (calculated isoelectric point = 4.85) with a calculated molecular mass of 22 kD.

The amino acid sequence was compared with the Swissprot





60 AAAATAATATTGAGATTGTTGAATGTGTTAAGTGCGGACATCAGATGCGAGAAGCAGACA . 120 AAGAAGCCCGCGATCACGTTCGCAAAGATGAGCAAGTGATCGGGTTTTTATCCGGACTAG -35 -10. 180 CGATATGCGCCGTGTTTTTTTAAGCTAGTG<u>AGTACA</u>CGGCTGCAGGAATTCCGC<u>TACAAT</u> 240 rbs. CTGCGCCACTATTCTTCCCATGCTCAGGAGATATCATGAAAGTAGCAAAAGACCTGGTGG . 300 S 360 GTGCGCCGCTGGACTACCTGCATGGTCACGGTTCCCTGATCTCTGGCCTGGAAACGGCGC G G G

FIG. 1. Nucleotide and amino acid sequence of WHP. A putative ribosome binding site (Scherer *et al.*, 1980; Stormo *et al.*, 1982) is *underlined* and indicated (*rbs*). A possible promoter sequence (Harley and Reynolds, 1987) has been defined by homology (indicated by -35 and -10), and the sequence of a putative terminator, found by computer search (Brendel and Trifonov, 1984), is indicated by *two arrows*. The beginning of the second domain is indicated. The peptide sequences determined by Edman degradation are *underlined*.

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ment cloned after the first PCR obviously only encodes the major  $NH_2$ -terminal degradation product of WHP. The sequence of the  $NH_2$ -terminal degradation product, as defined by the two tryptic peptides, exactly coincides with the region of homology between WHP and the two ORFs, suggesting that the  $NH_2$  terminus is a distinct domain.

The COOH-terminal part of WHP contains a cluster of potentially metal-chelating amino acids such as histidine, cysteine, and aspartic and glutamic acid, suggesting that one or several metal ions are bound there. Since WHP is a cytoplasmic protein the cysteines will be in the reduced form and can potentially take part in chelating metal ions. A direct internal repeat (Fig. 2a) within this domain can be noted. The sequence of the COOH terminus of WHP allows for the possibility that different metal ions are bound by different residues of WHP, according to their coordination requirements.

Overexpression and Purification of Recombinant WHP-To characterize WHP further, an overexpression system was constructed. Since WHP is a soluble cytoplasmic protein, a direct expression system using a strong repressible promotor seemed to be the obvious choice. The gene encoding WHP was cloned into the vector pUHE25-2.<sup>2</sup> The vector provides the very strong A1 phage promoter made repressible by the introduction of lac repressor binding sites. The gene was inserted right after the phage T5-derived ribosomal binding site II provided by the vector, resulting in the plasmid pWHPexpr. RB791 (Brent and Ptashne, 1981), a LacI-overproducing E. coli W3110 derivative, was used as expression strain. The recombinant protein was expressed with a yield of 50 mg/liter culture. It was found to remain soluble, and no deleterious effect on the cells was noted. The protein was purified by IMAC on a Ni<sup>2+</sup>-NTA column under native conditions as de2898

b

WHP

# Metal-binding E. coli Protein

A

G

G

a																	
147									L	A	H	G	H	v	Н	G	
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173	Ĥ	D	Ĥ	G	Ĥ	E	H	G	G	Е	Ġ	Ċ	ċ	Ġ	Ġ	K	
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(Table I). This indicates that metal ions can most efficiently be replaced via a direct exchange process. The metal binding domain of WHP thus seems to have a very strong need for a bound metal ion, but does not seem to care very much about the type of metal.

In the metal reconstitution experiments,  $Ni^{2+}$  and  $Zn^{2+}$  were the metals being bound best, but  $Cu^{2+}$  and  $Co^{2+}$  also bound to WHP, albeit to a smaller extent (Table I). It is noteworthy that  $Ni^{2+}$  and  $Zn^{2+}$  bind to WHP with 1:1 stoichiometry, leading to the conclusion that only one metal binding site is present in WHP despite the fact that a motif is present in direct repeat in the WHP metal binding domain. Iron did not bind to WHP at all, no matter whether it was added as FeIII or as FeII, complexed weakly by citrate or strongly by phenanthroline in the reconstitution experiment. Oxidation of FeII in slightly alkaline solution was prevented by addition of ascorbate.  $Mg^{2+}$  and  $Ca^{2+}$  did not bind to WHP under these conditions.

Comparing the metal reconstitution data to the data ob-

122	NHMLAGQNLKFNV <u>E</u> VVA	IREA
		22

147 T<u>EEELAHGHVAPAHDHHHDHDHDGCCGGHGHDHGHE</u>HGG<u>E</u>GCCGGKGNGGCGCH

1 MCTVCGCGTSAIEGHTHEVGDDGHGHHHHDGHHDHDHDHDHDHRGDHEHDDHHHA

54 <u>ED</u>GSVHYSKGIAGVHVPGMSQERIIQ...

#### Hydrogenase operon ORF

FIG. 2. a, special features of the metal binding domain of WHP. A direct repeat of 17 amino acids (amino acids 156–172 and amino acids 173–189) is *highlighted*. Identical amino acids are labeled with *vertical bars* and similar ones with *colons*. The two repeats contain 67% potentially metal ion binding amino acids. b, sequences of the COOH terminus of WHP and the NH<sub>2</sub> terminus of the translation product of an ORF from the hydrogenase operon from R. *leguminosarum* are shown in comparison in the *center lines*. Adjacent sequences of the respective proteins are shown in the *outer lines*. Potentially metal-chelating amino acids are *highlighted*.

scribed under "Material and Methods." Under these conditions, however, other host proteins still slightly contaminate the WHP fractions after IMAC, and a DEAE ion exchange chromatography step was used to obtain pure protein (Fig. 3). It is with this protein that the metal binding reconstitution studies were performed. As an alternative, in order not to introduce  $Ni^{2+}$ during the purification, WHP was also purified to homogeneity by a combination of chromatography on a hydroxylapatite and a DEAE ion exchange column as described under "Material and Methods." Metal Binding Assays—In order to determine whether there are metal ions bound by WHP after isolation from E. coli, the metal content of WHP solutions was assayed by AAS. For this purpose, samples were taken at the end of the native purification procedure involving IMAC: they contained  $Ni^{2+}$  and  $Zn^{2+}$ . Whereas Ni<sup>2+</sup> could have, but not necessarily has been, introduced during Ni<sup>2+</sup> IMAC purification by a metal exchange process,  $Zn^{2+}$  must have been bound to WHP already in the cell. Similarly, samples were taken at the end of the purification procedure involving the hydroxylapatite column chromatography. In this case,  $Ni^{2+}$ ,  $Zn^{2+}$ , and  $Ca^{2+}$ , but no  $Mn^{2+}$ ,  $Co^{2+}$ , or Cu<sup>2+</sup>, were found to be bound to WHP (data not shown). Using a similar argument as above  $Ca^{2+}$  may have been introduced by the hydroxylapatite column, whereas  $Zn^{2+}$  and  $Ni^{2+}$  should have been bound to WHP already in vivo. To determine which metal ion *can* bind to WHP, the protein, purified using IMAC, was stripped of metals by dialysis against an EDTA containing buffer, physiological in pH and salt concentration, and then incubated with a large molar excess of the metal ion to be tested. Ni<sup>2+</sup> and Zn<sup>2+</sup> both could not be removed completely by dialysis against EDTA, whereas subsequent incubation with a large molar excess of other divalent metal ions lead to a complete removal of  $Ni^{2+}$  and  $Zn^{2+}$  in all cases assayed tained by assaying WHP solutions at the end of the respective purification procedures, it is found that the results generally coincide well. Both kinds of experiments support the idea that  $Ni^{2+}$  and  $Zn^{2+}$  are preferentially bound. Yet in contrast to the metal reconstitution experiments, some  $Ca^{2+}$  was found to be bound to WHP when purified using the hydroxylapatite column. The high  $Ca^{2+}$  concentrations on the column seem to force WHP to accept the less preferred metal ion as a ligand. This again suggests that it is most important for WHP to have a metal ion bound at all.

In conclusion, the metal binding domain of WHP seems to be able to bind divalent metal ions within a certain range of ionic radii. Copper is bound to a smaller extent probably due to its strict geometrical coordination requirements. Iron is not bound at all, perhaps partly because of its small effective concentration in neutral or alkaline solutions due to oxo-bridging, and the affinity of WHP appears to be insufficient to sequester this metal. As  $Mg^{2+}$  is smaller and  $Ca^{2+}$  is larger than the divalent transition metals assayed, these metals do not seem to fit very well into the WHP metal binding site. Although a large number of oxygen ligands, preferred by these metals, are present in this domain, these metals bind only under extreme conditions or not at all. Homology Search—The NH<sub>2</sub> terminus of WHP and two homologous uncharacterized open reading frames, one in  $E. \ coli$ , one in P. fluorescens, have about 30% identity compared with each other (Fig. 4). Using these three sequences, a homology profile was constructed, and the Swissprot data base was searched with this profile. A significant homology to, and only to, the FKBP family showed up. Conversely, the three-dimensional homology profile of the FK506-binding protein (van Duyne et al., 1991) was calculated according to the method of Eisenberg and co-workers (Lüthy et al., 1992). This type of profile contains no sequence information at all, but is a measure of the compatibility of a sequence with a given three-dimensional structure. Searching the Swissprot data base with this profile, both the hypothetical 16.1-kDa protein of E. coli and the 16.3-kDa hypothetical protein of P. fluorescens were found with 8.4 and 7.1 standard deviations, second only to the FKBP family. Next a cluster alignment of the WHP, the 16.1-kDa and the 16.3-kDa protein and the bovine FKBP sequence, was performed (Fig. 4). It appears as if there might be an insertion in the structure of FKBP (Figs. 4 and 5). To obtain some initial experimental information about the protein structure, CD spectra were recorded (Fig. 6). They suggest a content of  $\beta$ -sheet of 51 ± 8% and 10 ± 4%  $\alpha$ -helix, and this would be consistent with a chain fold similar to FKBP, albeit with the uncertainty of the contribution of the large

FIG. 3. **Purification of WHP.** Shown is an SDS-PAGE gel of different fractions of the DEAE ion exchange column, used as the last step in the native purification procedure. The fraction number from a salt gradient is given below each lane. *M* denotes the molecular weight marker.



#### M 24 26 28 30 32 34 36 38 39 40 41 42 44 46

#### TABLE I Metal binding assays

WHP was stripped of its metal ions and was then reconstituted by incubation with a solution containing the metal ion specified in the first column. "EDTA" denotes that the metal was stripped from WHP without later metal ion addition, "none" denotes WHP receiving no treatment at all. The WHP samples treated according to the first column were assayed for their metal content by AAS, the type of metal ion being assayed is given in each column heading. Numbers give micromolar concentrations; ND, no determination. The numbers in the last column give the micromolar concentrations of WHP in the respective AAS samples.

<b>—</b>		WHP								
Treatment	Mg <sup>2+</sup>	Zn <sup>2+</sup>	Cu <sup>2+</sup>	Ni <sup>2+</sup>	Co <sup>2+</sup>	Co <sup>2+</sup> Fe <sup>2+</sup>		Ca <sup>2+</sup>	concentration	
				μ	м				μм	
$Mg^{2+}$	0	ND	ND	ND	ND	ND	ND	ND	52	
$Zn^{2+}$	ND	32	0	0	0	ND	ND	ND	30	
Cu <sup>2+</sup>	ND	0	37	0	0	ND	ND	ND	57	
Ni <sup>2+</sup>	ND	0	0	61	0	ND	ND	ND	57	
Co <sup>2+</sup>	ND	0	0	0	15	ND	ND	ND	54	
Fe <sup>2+</sup>	ND	ND	ND	ND	ND	0	ND	ND	47	
Fe <sup>3+</sup>	ND	ND	ND	ND	ND	ND	0	ND	40	

Ca <sup>2+</sup>	ND	0	88							
EDTA	0	4	0	19	0	0	0	0	130	
None	0	21	0	59	0	ND	ND	0	215	

insertion and the metal binding domain. In order to investigate the possibility that the predicted fold, similar to the FKBP family, allows for rotamase activity, a standard proline *cistrans*-isomerase activity assay (Liu *et al.*, 1990) was performed. No activity was found for WHP. It should be kept in mind, however, that the rotamase activity may be specific for a different type of substrate, and this cannot be excluded from this experiment.

Searching the data base with the COOH-terminal fragment and different amino acid patterns thereof led to the 50  $NH_2$ terminal amino acids of an open reading frame from the hydrogenase operon of Rhizobium leguminosarum (EMBL accession number: X52974) which is also rich in acidic amino acids, cysteine, and histidine (Hidalgo et al., 1992) (Fig. 2b). This ORF has three close homologs (PIR international data base accession numbers S23440, S15198, D38532) which only differ in the respective NH<sub>2</sub>-terminal parts (data not shown), suggesting that the  $NH_2$  termini are variable extensions of a common fold. Although no homology was found which was significant enough to provide a structural model for the COOH-terminal part of WHP, the analogy to the ORF from the hydrogenase operon suggests that such NH<sub>2</sub>- or COOH-terminal clusters of potentially metal-binding amino acids may form a separate functional domain.

#### DISCUSSION

WHP presumably is a two-domain protein, evidenced by homology and the facile proteolytic degradation at the putative domain border (see above). The NH<sub>2</sub>-terminal domain seems to contain an FKBP folding motif. This is suggested by three lines of evidence. First, sequence homology searches with either the NH<sub>2</sub>-terminal domain of WHP, or with the translation products of the homologous ORFs from E. coli and P. fluorescens, or with a sequence profile of the three aligned proteins shows that WHP and the two homologous ORFs are related to the FKBP family. Second, searching the data base with a FKBP structural profile (Lüthy et al., 1992), not containing any sequence information, retrieves the two ORFs, WHP not being in the data base, as the best structures directly after FKBPs at about 7 and 8.5 standard deviations. Third, taking a cluster alignment of WHP, the ORFs, and FKBP based on sequence homology, the 50 amino acids by which the NH<sub>2</sub>-terminal domain of WHP and its homologs are longer than FKBP are placed in a large loop of the FKBP structure where they can be accommodated (Figs. 4 and 5). The COOH-terminal metal binding motif of WHP is remarkable because it contains a cluster of almost 30 potentially metal-binding amino acids. Experimentally, the binding of

Similarity to WHP	*	+	+	*	+ -	+ ++	++*	+	+	+	*	+	*	++*	* + *	
WHP	MP	(VA	KDL	vvs	LA	YQVR	TEDG	VLVD	ESP	VSA	-PL	DYL	HGH	GSL	ISGLE	46
P. fluorescens	MTDQVLAEQF	RIC	GONT	EVI	LH	FALR	LENG	DTVD	STF	DKA	-PA	TFK	VGD	GNL	LPGFE	54
E. coli	MSE	5VÇ	SNS	AVL	VH	FTLK	LDDG	TTAE	STR	NNG	KPA	LFR	LGD	ASL	SEGLE	49
FKBP	GVQVETISPGDO	RTFP	RGQ	TCV	VH	YTGM	LEDG	KKFD	SSR	DRN	KPF	KFM	LGK	QEV	IRGWE	60
FKBP structure	βββββββ			βββ	3ββ	ββββ	3	ββββ	3						αααα	
Similarity to FKBP					+*	+	* * * *	*	*+		*	*	*	+	*+*	

P. fluorescens
E. coli
FKBP
FKBP structure
Similarity to FKBP

Similarity to WHP	+ *	***	+	*+*	+*	+	+	*	+	*	+	++	+	
WHP	TAM	EGHEVGI	OKFDVAV	GANDAYG	QYDENL	VQF	RVP	KDVF	GVDE	LQV	IG-	MRFI	LAETDQGP	v 10
P. fluorescens	AAL	FGFKAGI	OKRTLQI	LPENAFG	QPNPQN	VQI	IP	RSQF	NMDL	SE-	-GI	LVI	FNDAANTE	L 11:
E. coli	QHL	LGLKVGI	OKTTFSL	EPDAAFG	VPSPDL	IQY	YFS	RREF	DAGE	PEI	G	AIMLE	FTAMDGSE	м 109
FKBP	EGV	AQMSVG	RAKLTI	SPDYAYG.	ATGH									- 87
FKBP structure	ααα		ввввв	3										
Similarity to FKBP	+	**+	+ + +	++ *+*										

147

151

150

107

Similarity to WHP	* + + + + * ** *** + *++*++ +
WHP	<b>PVEITAVEDDHVVVDGNHMLAGQNLKFNVEVVAIREATEEE</b>
P. fluorescens	PGVVKAFDDAQVTIDFNHPLAGKTLTFDVEIIDVKAL
E. coli	PGVIREINGDSITVDFNHPLAGOTVHFDIEVLEIDPALEA-
FKBP	PGIIPPHATLVFDVELLKLE
FKBP structure	βββββββββββββββββββββββββββββββββββββββ

Similarity to FKBP

FIG. 4. Sequence alignment of WHP, the homologous ORFs from E. coli (Swissprot accession number P22563) and P. fluorescens (Swissprot accession number P21863), and the human FK506-binding protein (Brookhaven data base entry 1FKF), as performed by **CLUSTAL.** The line on top of the WHP sequence shows positions at which the first three proteins of the alignment are similar (+) or identical (\*). For comparison, the amino acids being identical or similar between all four aligned proteins are marked in the last line denoted "Similarity to FKBP." A star signifies an amino acid that is identical between FKBP and at least two of the aligned proteins; a plus denotes 4 similar amino acids. The line called "*FKBP structure*" shows the amino acids being involved in elements of secondary structure as determined by x-ray crystallography.  $\alpha$  denotes  $\alpha$ -helix, and  $\beta$  denotes  $\beta$ -sheet as the respective secondary structure element.



FIG. 5. Proposed overall structure of WHP. The structure is based on the human FK506-binding protein structure (van Duyne et al., 1991; Brookhaven protein data base entry 1FKF) and drawn with MOL-SCRIPT (Kraulis, 1991). The box on top shows where the large insertion of WHP and the two homologous translation products of the ORFs, compared with FKBP, would lie in the structure. The independent COOH-terminal metal binding motif is represented by the box on the right.

transition metal ions has been shown. For Zn<sup>2+</sup> and Ni<sup>2+</sup>, which are carried through at least one complete purification procedure, it is probable that they are bound *in vivo*. Judging from the metal binding experiments,  $Ni^{2+}$  and  $Zn^{2+}$  have the highest binding constant of all metals tested. To conclusively answer the question which metals are bound in vivo, the binding constants of different metals have now to be determined. Usually divalent metal ions are coordinated by a finite small number of chelating amino acids often distant in sequence (Glusker, 1991; Christianson, 1991). In contrast, the COOHterminal domain of WHP essentially consists of a stretch of

260

potentially metal-binding amino acids: 28 out of 50 amino acids can potentially take part in chelating metal ions. This is reflected by the ability of WHP to bind different metals such as Ni<sup>2+</sup>, Zn<sup>2+</sup>, Co<sup>2+</sup>, Cu<sup>2+</sup>, and Ca<sup>2+</sup>, at least to some extent. Being flexible enough to accommodate metal ions with rather different coordination requirements, WHP might not be able to use the bound metal ion as a permanent structural anchor. It rather seems possible that the COOH-terminal part might exist in several energetically similar structures, depending on the nature of the metal ion bound and its preferred coordination. Since the COOH-terminal motif is devoid of hydrophobic amino acids, it is questionable whether the domain has a defined intrinsic fold. In contrast, the binding of a metal ion might only induce a defined structure, making the motif a sensor. The COOH-terminal part of WHP is rich in histidines in close proximity of acidic amino acids, which might shift the  $pK_a$  of the histidines to more alkaline values. The actual charge of the histidines, and thereby their chelating ability, may therefore depend on small pH changes in the physiological range. It is suggested that the motif functions as a metal or proton sensor by temporary metal ion binding inducing a specific fold. This in turn might either influence the activity of the NH<sub>2</sub>-terminal domain or the interaction with other proteins.

Does WHP fulfill this putative task in an unbiased fashion regarding different divalent metal ions? WHP binds different divalent metal ions, preferentially  $Ni^{2+}$  and  $Zn^{2+}$ . Whereas Zn<sup>2+</sup>-binding proteins are ubiquitous, only four classes of Nibinding enzymes have been described so far: hydrogenases (Hausinger, 1987), CO-dehydrogenases (Conover et al., 1990), methyl-coenzyme M-reductases (Zimmer and Crabtree, 1990) and ureases (Dixon et al., 1975). Of these enzymes, only hydrogenases are known to occur in E. coli. The oxidation of molecular hydrogen is nickel-dependent (Evans et al., 1987). Three hydrogenases have been discovered in E. coli, and the genes are each arranged in operons with other genes of unknown functions (Menon et al., 1990; Böhm et al., 1990; Przybyla et al., 1992). Another operon contains functions essential for all three hydrogenases (Lutz et al., 1991). The existence of a cluster of metal-binding amino acids (ORF in the hydrogenase operon of R. leguminosarum whose  $NH_2$  terminus is similar to the COOH terminus of WHP) in a protein, probably with an accessory role in hydrogenase activity, makes it tempting to speculate that WHP might also be involved in related functions. Since WHP is abundant in the cell and the gene of WHP is not part of an operon, however, as it seems to have a promoter and terminator flanking it, a more general role of this protein is probable. In summary, it is suggested that the COOH-terminal part of WHP acts as a sensor modulating the function of the  $NH_2$ terminal domain, which may be a substrate-specific rotamase. The COOH terminus may respond to more general changes in physiology by binding  $Zn^{2+}$ , pH-dependent or not, or to a more specific situation binding  $Ni^{2+}$  in vivo. In addition, WHP was found as the product of a gene required for phage lysis (termed slyD) (Roof et al., 1993). In this case WHP is necessary for the phage gene E-mediated cell lysis to occur. Even while the cellular function of WHP has not yet been established, it may deserve some further investigation. First, the determination of the structure of the  $NH_2$ -terminal domain may test the hypothesis of the FKBP fold and lead to valuable insight how the FKBP structure as a module may be extended by enlarging a loop. Second, the small size of the COOH-terminal part may make it a useful module to confer general divalent metal ion binding ability with a high binding constant to any protein of interest. Third, nuclear and electronic magnetic resonance, as well as extended x-ray absorption fine structure spectroscopy may be used to gain insight into the way various cations are bound. Particularly interesting is the question whether different metals can choose their own coordinating residues from the large offering of this motif. These investigations may be especially useful for metals for which only few natural models are available, such as Ni<sup>2+</sup>. Finally, it is unresolved whether a motif with essentially no hydrophobic residues has a defined structure at all. In conclusion, our evidence suggests that WHP is a twodomain protein being composed of one structurally rigid domain with a fold partially similar to FKBP, and one flexible motif being able to bind metal ions, possibly as a means of modulating the function of the NH<sub>2</sub>-terminal domain, in response to  $Zn^{2+}$ ,  $Ni^{2+}$ , or pH.

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